

Synthesis and NK₁/NK₂ Receptor Activity of Substituted-4(*Z*)-(methoxyimino)pentyl-1-piperazines

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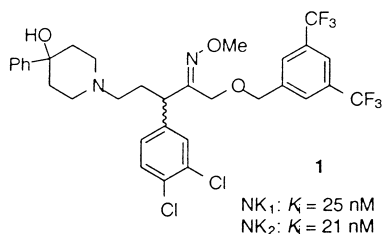
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Abstract—A series of 5-[(3,5-bis(trifluoromethyl)phenyl)methoxy]-3-(3,4-dichlorophenyl)-4(*Z*)-(methoxyimino)pentyl-1-piperazines was prepared and their affinity for the NK₁ and NK₂ receptors investigated. Compounds **7f**, **10o**, **10r**, and **10s** were found to be our most potent inhibitors. © 2000 Elsevier Science Ltd. All rights reserved.

Introduction

Substance P (SP), neurokinin A (NKA), and neurokinin B (NKB) are members of a family of neuropeptides that share a common C-terminal sequence of Phe-X-Gly-Leu-Met-NH₂ and are found throughout the central and peripheral nervous systems.¹ These peptides exert their biological effect through three neurokinin receptors (NK₁, NK₂, and NK₃), which are each G-protein linked to secondary messenger systems. Although each neurokinin can act as an agonist at all three receptor subtypes, substance P (SP), neurokinin A (NKA), and neurokinin B (NKB) have the highest affinity for the NK₁, NK₂, and NK₃ receptor, respectively.² The neurokinins may play an important role in a number of disease states including migraine, emesis, pain, arthritis, depression, anxiety, and asthma.³ In particular, it is believed that both SP and NKA are responsible for the excessive mucus secretion, airway constriction, and plasma extravasation found in the pathology of asthma.⁴ Therefore, our goal was to discover a compound which could block both the NK₁ and NK₂ receptors as a therapy for the treatment of asthma.



Biological screening⁵ of directed targets and synthetic intermediates in the program identified oxime **1** to be a potent dual NK₁ and NK₂ receptor antagonist.⁶ In this communication, we describe a series of analogues in which the piperidine ring has been replaced by an array of substituted piperazines.

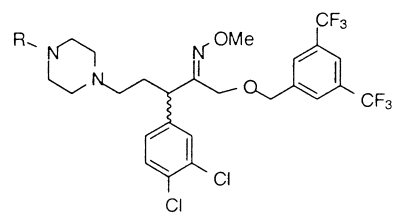
Results and Discussion

The target piperazines **7a–f** were prepared as the racemates by the synthetic route shown in Scheme 1.⁷ 1,4-Addition of ethyl 1,3-dithiolane-2-carboxylate (**2**) to methyl *trans*-3,4-dichlorocinnamate gave the diester **3**.⁸ Reduction of the diester **3** with lithium aluminum hydride and selective protection of the less hindered primary alcohol afforded the *t*-butyldimethylsilyl ether **4**. Alkylation of alcohol **4** with 3,5-bis(trifluoromethyl)benzyl bromide and removal of the silyl protecting group yielded compound **5**. Hydrolysis of the dithiane **5**,⁹ treatment with methoxylamine,¹⁰ and Swern oxidation¹¹ provided aldehyde **6**. Reductive amination¹² with aldehyde **6** produced the piperazines **7a–f**.

The biological activity of piperazines **7a–f** are summarized in Table 1. These results indicate that the unsubstituted piperazine **7a**, the alkylsubstituted piperazines **7b** and **7c**, and the arylsubstituted piperazines **7d** and **7e** all bind to the NK₁ receptor in favor of the NK₂ receptor. In this set of analogues, only the (pyrrolidinocarbonylmethyl)piperazine **7f** stands out as a dual NK₁/NK₂ antagonist.

Additional analogues based on the (pyrrolidinocarbonylmethyl)piperazine lead structure **7f** are reported in

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Table 1. NK₁ and NK₂ antagonistic activity of the piperazines **7a–f**


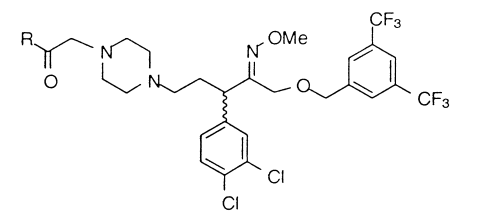
The chemical structure shows a piperazine ring substituted with an R group. The other end of the piperazine is connected via a methylene chain to a chiral center. This chiral center is also bonded to a 2,4-dichlorophenyl group and an oxime group (-N=OMe). The oxime is further connected via a methylene group to a 3,5-bis(trifluoromethyl)benzyl ether group.

Compd	R	NK ₁ <i>K_i</i> (nM) ^a	NK ₂ <i>K_i</i> (nM) ^a
7a	H	30	738
7b	Me	61	200
7c	Cyclohexyl	53	128
7d	Ph	181	978
7e	2-Pyrimidinyl	70	217
7f	<i>N</i> -PyrrolidinylCOCH ₂	14	21







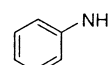

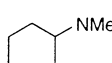

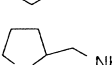


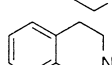

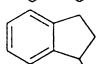

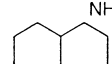


^aValues are means of two experiments.

Table 2. These compounds were prepared by reductive amination of key aldehyde **6** with TROC-protected piperazine. Removal of the TROC moiety with zinc under buffered pH conditions avoided oxime isomerization which was observed under acidic conditions.¹³ The analogues **10a–k** in Table 2 include a broad range of variation at the terminal amine position. Alkylation of piperazine **7a** in a parallel synthesis manner¹⁴ gave target analogues **10a–k**. Relative to the standard **7f**, none of the piperazine analogues **10a–k** shows improved potency as a dual NK₁/NK₂ antagonist.

The analogues **10m–w** focus on a more specific structure–activity study of **7f**. Replacement of the pyrrolidine ring **7f** with the smaller azetidine ring as in **10m** or **10n** retains both NK₁ and NK₂ potency. The 3-position of pyrrolidine can be substituted with a hydroxy moiety as

Table 2. NK₁ and NK₂ antagonistic activity of the piperazines **10a–w**


The chemical structure shows a piperazine ring substituted with an R group. The other end of the piperazine is connected via a methylene chain to a chiral center. This chiral center is also bonded to a 2,4-dichlorophenyl group and an oxime group (-N=OMe). The oxime is further connected via a methylene group to a 3,5-bis(trifluoromethyl)benzyl ether group.

Compd	R	NK ₁ <i>K_i</i> (nM)	NK ₂ <i>K_i</i> (nM)	Compd	R	NK ₁ <i>K_i</i> (nM) ^a	NK ₂ <i>K_i</i> (nM) ^a
10a		82	208	7f		14 (±7)	21 (±7)
10b		62	259	10m		23 (±5)	33 (±8)
10c	(iPr) ₂ N	82	170	10n		22 (±5)	31 (±3)
10d	EtOCH ₂ CH ₂ CH ₂ NH	39	382	10o		16 (±4)	30 (±10)
10e		125	>680	10p		9 (±2)	48 (±25)
10f		113	368	10q		14 (±2)	95 (±26)
10g		20	266	10r		14 (±4)	13 (±2)
10h	EtOOCCH ₂ CH ₂ N	24	409	10s		13 (±4)	24 (±6)
10i		125	566	10t		24 (±6)	26 (±8)
10j		136	>680	10u		13 (±3)	38 (±21)
10k		136	477	10v		47 (±24)	52 (±8)
				10w		19 (±4)	28 (±6)

^aValues are the means of three experiments; standard deviation is given in parentheses.

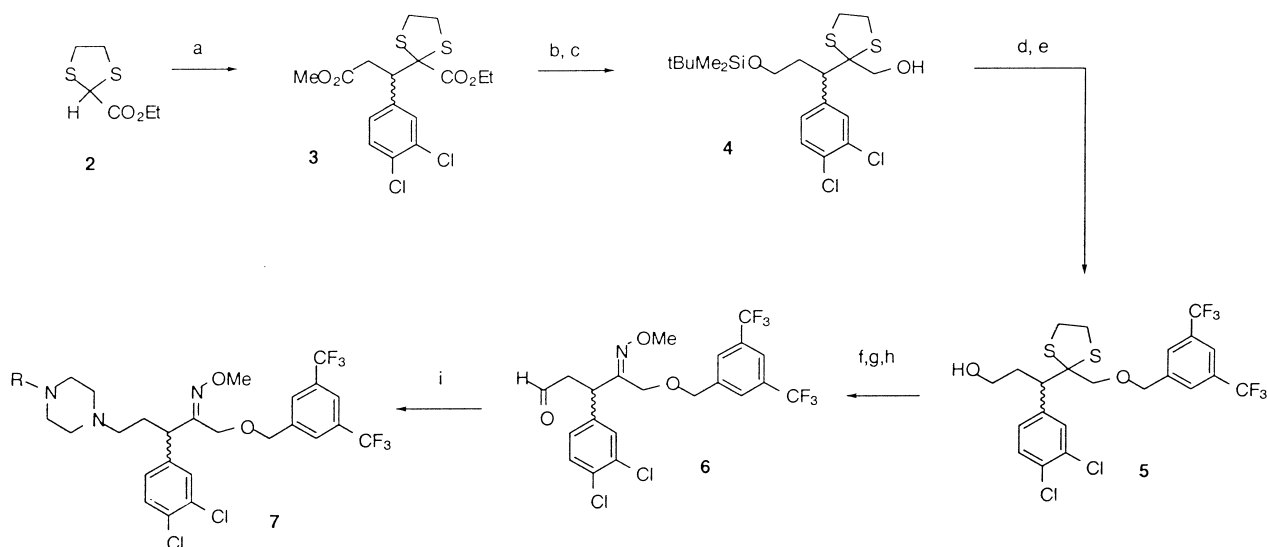


Figure 1. (a) $\text{LiN}(\text{TMS})_2$, THF, -78°C , methyl *trans*-3,4-dichlorocinnamate, 87%; (b) LiAlH_4 , THF, 0°C – 23°C , 92%; (c) *t*-BuMe₂SiCl, Et₃N, DMAP, THF, 0°C – 23°C , 96%; (d) $\text{KN}(\text{TMS})_2$, THF, 3,5-bis(trifluoromethyl)benzyl bromide, 0°C – 23°C , 88%; (e) HF, CH₃CN, 100%; (f) $\text{Hg}(\text{ClO}_4)_2$, CaCO₃, THF, H₂O, 100%; (g) MeONH₂·HCl, NaOAc, EtOH, H₂O, Δ , 58% of *Z* isomer and 25% of *E* isomer; (h) ClCOCOCI, DMSO, CH₂Cl₂, Et₃N, 100%; (i) substituted piperazine, NaCNBH₃, CF₃CH₂OH, 3 Å sieves, 20–40%.

in **10o** and maintain biological activity. However, the larger acetamide group as in **10p** or the dimethylamino group as in **10q** both tend to decrease NK₂ potency but not NK₁ potency. The 2-position of the pyrrolidine is tolerant to substitution by a hydroxymethyl moiety as in the stereoisomers **10r** and **10s** or a carboxamide moiety as in **10t** and retains the dual neurokinin antagonist profile. In comparison to the hydroxypiperidine analogue **10u**, the hydroxymethylpiperidine **10v** shows a slight decrease in both NK₁ and NK₂ potency. The morpholine analogue **10w** possesses potent dual NK₁/NK₂ activity.

Conclusion

In conclusion, we have found that the 4-hydroxy-4-phenylpiperidine ring of compound **1** can be replaced by a totally different structural subunit—the (pyrrolidinocarbonylmethyl)piperazine **7f**. Modification of the pyrrolidine ring of **7f** with a hydroxy group retains biological activity, and analogues such as **10o**, **10r**, and **10s** are equipotent to **1** as a dual NK₁/NK₂ antagonist. Further structure–activity relationship studies will be reported in future publications.

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References and Notes

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- Our parallel synthesis procedure was to react 11 different amines with bromoacetyl bromide in separate vials and subsequently treat with piperazine **7a**. The reaction mixtures underwent water work up, extraction, and analysis by mass spectroscopy.